

**TOTAL SYNTHESIS OF BIOLOGICALLY IMPORTANT AMINO SUGARS VIA
 THE NITROALDOL REACTION**

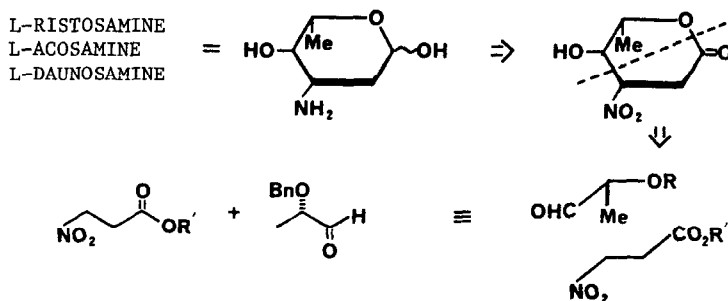
Stephen Hanessian* and John Kloss
 Department of Chemistry, Université de Montréal
 Montréal, Québec, Canada H3C 3V1

Summary. Aldol condensation between *O*-benzyl D- or L-lactaldehyde and alkyl 3-nitropropionates in the presence of various catalysts leads to stereoisomeric nitroaldol products which can be transformed into aminodeoxy hexoses in the D- or L-series.

The amino sugars occupy a unique role in natural product chemistry, either as components of antibiotics¹ or biopolymers.² The need to develop novel and efficient syntheses of isomeric 3-amino-2,3,6-trideoxy-L-hexoses, which are components of the anthracycline antitumor agents³ and other natural products is evident, since experience has shown that even minor stereochemical or functional variations can result in dramatic changes in their biological profile.^{3,4} Traditionally, most of such syntheses started with carbohydrate precursors and proceeded through the systematic modification of existing functional groups.⁵ More recently, imaginative approaches from non-sugar precursors have been reported.⁶

We now describe a versatile and novel route to 3-amino-2,3,6-trideoxy-D- or L-hexoses by a bond-forming strategy which engages a suitably protected D-⁷ or L-lactaldehyde¹³ and an alkyl 3-nitropropionate in a stereocontrolled nitroaldol reaction^{8,9} (Scheme 1). The naturally occurring L-acosamine,¹⁰ L-daunosamine,^{3,11} and L-ristosamine¹² are thus accessible by this methodology and by further manipulation of the condensation products.

Scheme 1

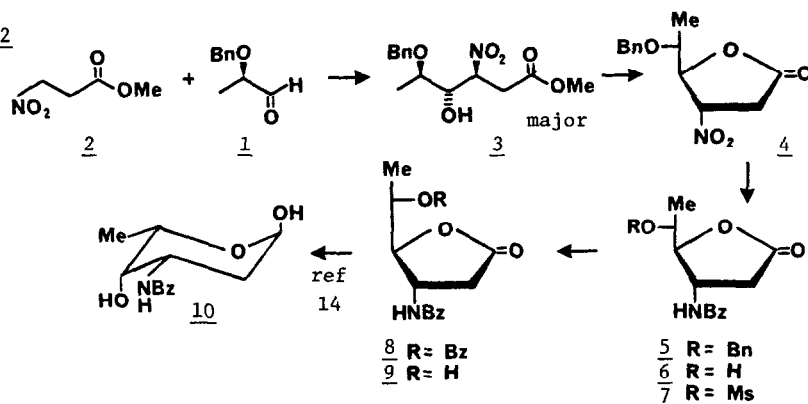


(*R*)-*O*-Benzyl ethyl lactate,^{7,13} $\alpha_D +76.5^\circ$ ($l=1$); was carefully reduced (DIBAL, Et₂O, -78°C) to give *O*-benzyl-D-lactaldehyde **1**, $\alpha_D +61.5^\circ$ ($l=1$). Treatment of **1** with methyl 3-nitropropionate **2** (0.84 equiv.) in the presence of neutral alumina⁹ (0.4 g/mmmole, 48h), led to a mixture of three nitroaldol products from which the preponderant crystalline D-ribo isomer **3** was isolated by direct crystallization and by subsequent chromatography of the mother liquor (62%); m.p. 108-109°C, $[\alpha]_D -33.6^\circ$ (cl.06, CH₂Cl₂). Interestingly, the

D-ribo isomer was the thermodynamic product in the reaction (ribo:xylo:arabino, 15, 1½, 1), since shorter reaction times gave a lower stereoselection (Scheme 2). Lactonization of 3 (HCl/CH₂Cl₂) gave the desired γ -lactone 4 in quantitative yield; $[\alpha]_D -40.9^\circ$ (c1.39, CH₂Cl₂). Reduction of the nitro group (Ra-Ni 500% w/w, Bz₂O, MeOH) with in situ N-benzoylation gave 5 in 71% yield; m.p. 132-133°C, $[\alpha]_D -39.4^\circ$ (c1.08, CH₂Cl₂). Hydrogenolysis (5% Pd/C, MeOH, H⁺) gave 6 in quantitative yield; m.p. 145-148°C, $[\alpha]_D -43.1^\circ$ (c1.13, EtOH). The D-ribo lactone 6 was subjected to configurational inversion at C-5 to give the benzoate 8, (NaOBz, DMF, 60%); $[\alpha]_D +37.5^\circ$ (c0.8, CH₂Cl₂) via mesylate 7, (MsCl, pyr, 91%); m.p. 144-146°C, $[\alpha]_D -17.7^\circ$ (c1.05, MeOH). Catalytic debenzoylation (cat. NaOMe/MeOH, H⁺, 79%) then gave the known L-lyxo lactone 9¹⁴; m.p. 138-140°C, $[\alpha]_D -24.9^\circ$ (c1, EtOH) which had been previously converted into N-benzoyl-L-daunosamine 10 by Fuganti and coworkers.¹⁴

It should be noted that whereas the (R)-D-lactaldehyde 1 gave the D-ribo lactone 6, the (S)-L-lactaldehyde 11 gave the L-ribo lactone 16.

Scheme 2



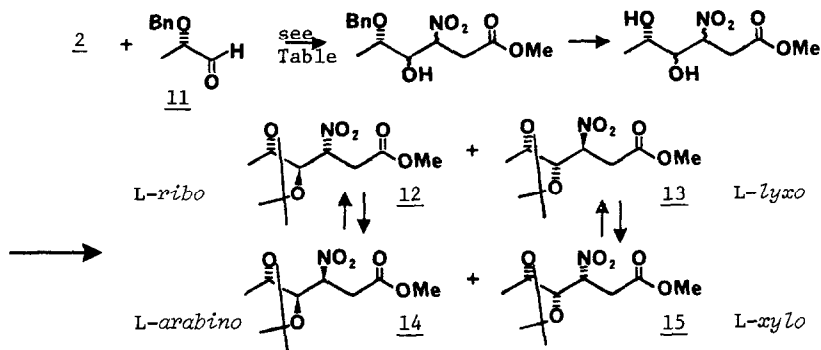
Access to immediate precursors of L-ristosamine and L-acosamine was possible by application of the nitroaldol reaction in the L-series analogously to that as described above, and in the presence of other catalysts. Thus, the mixture of nitroaldol condensation products from O-benzyl-L-lactaldehyde⁷ 11, $a_D -67.7^\circ$ ($l=1$), was debenzylated (5% Pd/C, MeOH, H⁺), then subjected to ketalization (acetone/dimethoxypropane, CSA), and chromatography to give the L-ribo acetonide 12 as the major product, (43%, three steps); $[\alpha]_D +70.4^\circ$ (c0.95, CH₂Cl₂). As in the D-series, only minor amounts of the L-xylo, 15 (4.5%, three steps), $[\alpha]_D -12.8^\circ$ (c1.18, CH₂Cl₂); the L-arabino, 14 (2.5%, three steps), $[\alpha]_D +58.0^\circ$ (c0.64, CH₂Cl₂); and the L-lyxo, 13 (trace), $[\alpha]_D +6.04^\circ$ (c1.07, CH₂Cl₂) isomers were formed.

It also was possible to effect equilibration at C-3 of the individual acetonides. Treatment of the L-arabino derivative 14 with tBuOK in THF (0.25 equiv., 24h, 0°C) led to a 3:7 mixture of 14 and the L-ribo isomer 12. Similar treatment of the L-xylo isomer 15 gave a 2:3 mixture of 15 and the L-lyxo isomer 13 (Scheme 3). These could be individually isolated by chromatography.

A nitroaldol condensation also was done in the presence of a catalytic amount of tBuOK (5%, THF, 0°, 2h). In this case, the L-arabino isomer was predominant, but the stereoselection was much lower as compared to alumina (Table 1).

Anticipating the possible influence of the cation on the stereochemistry of the nitroaldol condensation through the intermediacy of metal-coordinated transition states, the reaction was conducted in the presence of various salts of divalent metals. Thus, in the

Scheme 3



presence of $t\text{BuOK}/\text{ZnBr}_2$ (1 equiv. each, THF, 0° , 2.5h), the L-arabino isomer was favored, whereas with $t\text{BuOK}/\text{MgBr}_2$ (1 equiv. each, THF, 0° , 2.5h), there was a bias toward the L-xylo isomer.

Table 1

	<u>L-ribo</u>	<u>L-lyxo</u>	<u>L-xylo</u>	<u>L-arabino</u>
cat. $t\text{BuOK}$ (0°)	1	trace	1	2
$t\text{BuOK}$ (0°)	2	trace	1	1
$t\text{BuOK}/\text{MgBr}_2$ (0°)	1	trace	3	1
$t\text{BuOK}/\text{ZnBr}_2$ (0°)	1	trace	1	3
$t\text{BuOK}/\text{TMSCl}/\text{F}^-$ (-78°)	1	0	0	1
neutral alumina	15	trace	1.5	1

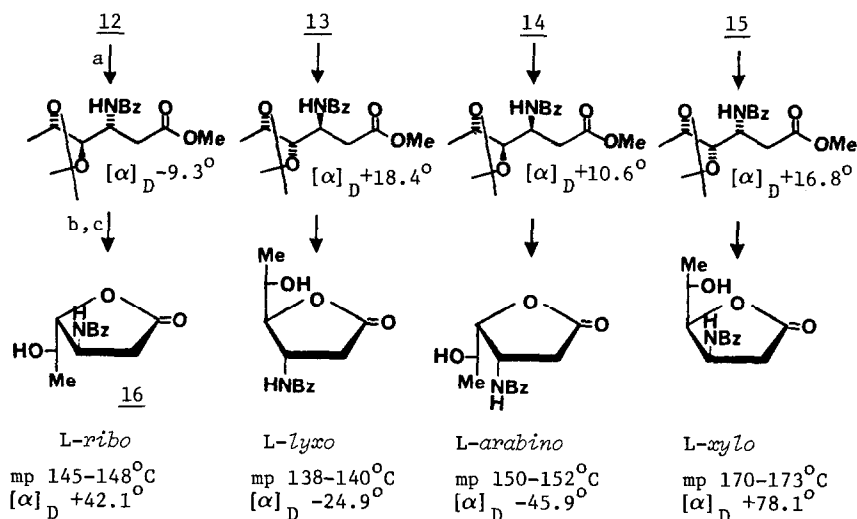
In order to test the effect of a chiral auxiliary function, the potassium *t*-butoxide condensation was done with the d- and l-menthyl 3-nitropropionates,¹⁵ and the reaction mixture processed to the acetonide stage. Surprisingly, with the d-ester, a 1:1:2 ratio (L-ribo:L-xylo:L-arabino) of isomers was obtained; while with the l-ester, the L-ribo and L-xylo isomers predominated (1:3 respectively). Poor stereoselection was observed with the racemic ester, indicating an inherent stereochemical bias with the optically active esters, rather than a purely steric effect. Conversely, condensations on alumina gave exactly the same ratios (1:1:1, ribo:xylo:arabino) when either the d- or the l-menthyl ester was used, which indicates a purely steric effect in this case.

With an efficient protocol to obtain the nitroaldol products in hand, it was possible to gain facile access to the corresponding known N-benzoyl lactones¹⁴ (and hence the free sugars¹⁴) from the individual acetonides as shown in Scheme 4.

The results described above, particularly with the metal-assisted and alumina catalyzed stereocontrolled reactions, demonstrate a novel approach to the synthesis of aminodeoxy sugars and the corresponding β -amino acids.

Acknowledgements. We thank the Natural Sciences and Engineering Research Council of Canada, Le Ministère de l'Éducation du Québec and Farmitalia-Carlo Erba for financial assistance. We also thank Dr. C. Fuganti (Polytechnic of Milan) for samples of the various N-benzoyl lactones.

Scheme 4



reagents: a) Ra-Ni, Bz₂O, MeOH, H₂, 65-82%. b) TFA/H₂O 9/1, 0°C. c) HCl/CH₂Cl₂, 71-85%.

References:

- See for example, S. Umezawa, MTP International Science Review, vol. 7, Carbohydrates, Organic Series Two, 1976, p. 149; S. Hanessian and T.H. Haskell, "The Carbohydrates", vol. IIA, 1970, p. 139.
- See for example: V. Sharon, "Complex Carbohydrates, Their Chemistry, Biosynthesis and Functions", Addison-Wesley Publishing Co., Reading Mass. 1975.
- F. Arcamone, "Doxorubicin, Anticancer Antibiotics", Academic Press, New York, NY, (1981).
- A. Suarato, S. Penco, A. Vigevani, and F. Arcamone, *Carbohydr. Res.*, **98**, C1 (1981).
- See for example, A. Bargiotti, G. Cassinelli, S. Penco, A. Vigevani, and F. Arcamone, *Carbohydr. Res.*, **100**, 273 (1982); T. Yamaguchi, and M. Kojima, *Carbohydr. Res.*, **59**, 343 (1977); D. Horton, and W. Weckerle, *Carbohydr. Res.*, **46**, 227 (1976), and references cited therein; B. Fraser-Reid, and H.W. Pauls, *J.C.S. Chem. Comm.*, 1031 (1983).
- See examples cited by G.J. McGarvey, M. Kimura, T. Oh, and J.M. Williams, *J. Carbohydr. Chem.*, **3** (2), 125-188 (1984); see also T. Hiyama, K. Nishide, and K. Kobayashi, *Tetrahedron Lett.*, **25**, 569 (1984); C. Fuganti, P. Grasselli, G. Pedrocchi-Fantoni, *J. Org. Chem.*, **48**, 909 (1983), and previous papers; for a synthesis of racemic compounds, see C.H. Heathcock, and S.M. Montgomery, *Tetrahedron Lett.*, **24**, 4637 (1983).
- Prepared from (S)-ethyl lactate using the procedure of L.E. Overman, K.L. Bell, and F. Ito, *J. Am. Chem. Soc.*, **106**, 4192 (1984) to give the enantiomeric (R)-ethyl lactate, then benzylation as in Ref. 13.
- D. Seebach, A.K. Beck, T. Mukhopadhyay, and E. Thomas, *Helv. Chim. Acta*, **65**, 1101 (1982).
- G. Rosini, R. Ballini, and P. Sorrenti, *Synthesis*, 1014 (1983).
- N.N. Lomakina, I.A. Spiridonova, Y.N. Sheinker, and T.F. Vlassova, *Khim. Prir, Soedin.*, **9**, 101 (1973); *Chem. Abstracts*, **78**, 148170m (1973).
- F. Arcamone, G. Cassinelli, P. Orezzi, G. Franceschi, and R. Mondelli, *J. Am. Chem. Soc.*, **86**, 5335 (1964).
- R. Bogнар, F. Sztaricskai, M.E. Munk, and J. Tamas, *J. Org. Chem.*, **39**, 2971 (1974).
- Benzylation of (S)-ethyl lactate (Aldrich $[\alpha]_D -12^\circ$ (neat)) was performed according to the procedure of K. Mislow, R.E. O'Brien, and H. Schaefer, *J. Am. Chem. Soc.*, **84**, 1940 (1962). See also S.J. Abbott, S.R. Jones, S.A. Weinman, F.M. Bockhoff, F.W. McLafferty, and J.R. Knowles, *J. Am. Chem. Soc.*, **101**, 4323 (1979).
- G. Fronza, C. Fuganti, and P. Grasselli, *J.C.S. Chem. Comm.*, 442 (1980).
- Prepared from the acid chloride and either d- or l-menthol to give the d-menthyl ester; m.p. 64-65°C, $[\alpha]_D +61.7^\circ$ (c5, EtOH) and the l-menthyl ester; m.p. 64-65°C, $[\alpha]_D -63.8$ (c5, EtOH) respectively.
- After the preparation of this manuscript, a similar nitroaldol strategy appeared in print. See T. Suami, K. Tadano, A. Suga, Y. Ueno, *J. Carbohydr. Chem.*, **3**, 429 (1984).

(Received in USA 26 November 1984)